

Name: \_\_\_\_\_

Date: \_\_\_\_\_

## Cell Transport WebQuest

### **Biomembranes I: Membrane Structure and Transport**

Go to [http://www.phschool.com/science/biology\\_place/biocoach/biomembrane1/intro.html](http://www.phschool.com/science/biology_place/biocoach/biomembrane1/intro.html)

#### *Concept 1: Membrane Structure*

Membranes consist of a \_\_\_\_\_ combined with a variety of \_\_\_\_\_ in a fluid \_\_\_\_\_ arrangement.

The surfaces of cell membranes are \_\_\_\_\_ (water-loving); the interiors are \_\_\_\_\_ (water-fearing).

Hydrophilic molecules tend to interact with \_\_\_\_\_ and each other.

Hydrophobic molecules \_\_\_\_\_ interaction with water and tend to interact with other \_\_\_\_\_ molecules.

#### *Concept 2: Osmosis*

Osmosis (movement of \_\_\_\_\_ across membranes) depends on the relative \_\_\_\_\_ of solute molecules on either side of the \_\_\_\_\_.

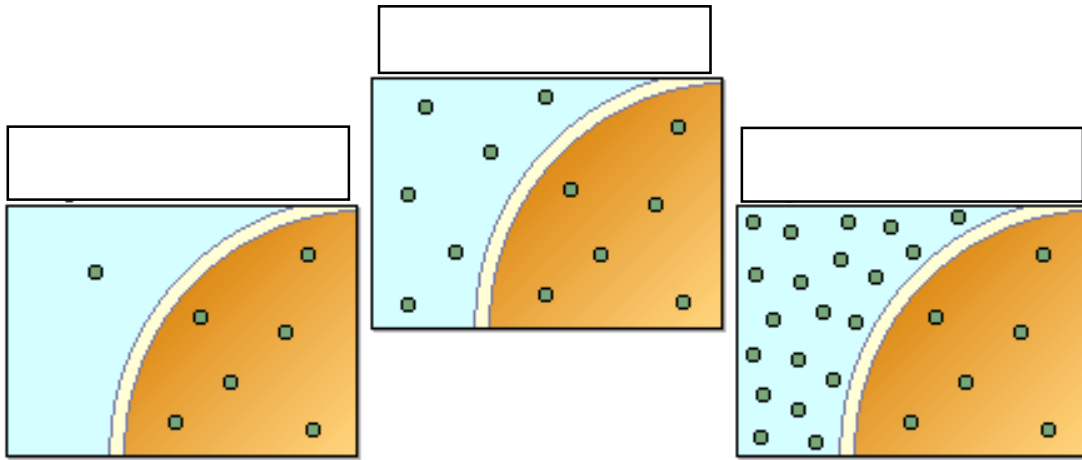
The presence or absence of cell \_\_\_\_\_ influences how cells respond to osmotic fluctuations in their environment.

**Click on "Review" in the upper right hand corner. It will open up a new screen.**

#### *Concept 2 Review: Isotonic, Hypotonic, and Hypertonic Solutions*

Water moves readily across cell membranes through special protein-lined \_\_\_\_\_, and if the total concentration of all dissolved solutes is not \_\_\_\_\_ on both sides, there will be net \_\_\_\_\_ of water molecules into or out of the cell. Whether there is net movement of water into or out of the cell and which \_\_\_\_\_ it moves depends on whether the cell's environment is isotonic, hypotonic, or hypertonic.

Label the three illustrations below as *isotonic*, *hypotonic*, or *hypertonic*.



Click "Next" on the bottom left side of the screen.

*Concept 2 Review: Cells in Isotonic Solutions*

When two environments are isotonic, the total concentration of dissolved solutes is the \_\_\_\_\_ in both of them.

When cells are in isotonic solution, movement of water out of the cell is exactly \_\_\_\_\_ by movement of water into the cell. A \_\_\_\_\_ solution of NaCl (saline) is isotonic to \_\_\_\_\_ cells.

Select "animate" to watch the movement of water molecules through protein channels in the cell membrane in an isotonic solution.

**Make your own sketch of a cell in an isotonic solution:**

Click "Next" on the bottom left side of the screen.

*Concept 2 Review: Cells in Hypotonic Solutions*

Hypotonic comes from the Greek "hypo," meaning \_\_\_\_\_ and "tonos" meaning \_\_\_\_\_. In a hypotonic solution, the total concentration of all dissolved solute particles is \_\_\_\_\_ than that of another solution or less than that of a \_\_\_\_\_.

If concentrations of dissolved solutes are less \_\_\_\_\_ the cell than \_\_\_\_\_, the concentration of water outside is correspondingly greater. When a cell is exposed to such hypotonic solutions, there is net movement of water \_\_\_\_\_ the cell. Cells without cell \_\_\_\_\_ will swell and may \_\_\_\_\_ (lyse) if excess water is not removed from the cell. Cells with cell walls often benefit from the \_\_\_\_\_ pressure that develops in hypotonic environments.

Select "animate" to watch the movement of water molecules across the cell in a hypotonic solution.

**Make your own sketch of a cell in a hypotonic solution:**

Select "A Closer Look" to see plant and animal cells in hypotonic environments on the bottom of the page.

**Compare and contrast what happens when a plant cell and animal cell are placed in a hypotonic solution.**

Click "Back to Review" and then select "Next" to view cells in hypertonic solutions.

*Concept 2 Review: Cells in Hypertonic Solutions*

Hypertonic comes from the Greek "hyper," meaning \_\_\_\_\_, and "tonos," meaning stretching. In a hypertonic solution, the total concentration of all dissolved solute particles is \_\_\_\_\_ than that of another solution, or greater than the concentration in a \_\_\_\_\_.

If concentrations of dissolved solutes are greater \_\_\_\_\_ the cell, the concentration of water outside is correspondingly \_\_\_\_\_. As a result, water inside the cell will flow \_\_\_\_\_ to attain equilibrium, causing the cell to \_\_\_\_\_. As cells lose water, they lose the ability to \_\_\_\_\_ or \_\_\_\_\_. Hypertonic environments, such as concentrated brines or \_\_\_\_\_, have been used for food preservation because microbial cells that would otherwise cause spoilage are \_\_\_\_\_ in these very hypertonic environments and are unable to function.

Select "animate" to watch the movement of water molecules across the cell in a hypertonic solution.

**Make your own sketch of a cell in a hypertonic solution:**

Select "A Closer Look" to see plant and animal cells in hypertonic environments on the bottom of the page.

**Compare and contrast what happens when a plant cell and animal cell are placed in a hypertonic solution.**

Close the review screen you are on to go back to the BioCoach Activity. You should still be on *Concept 2: Osmosis*. Click “Next Concept.”

### ***Concept 3: Selective Permeability of Membranes***

Cell membranes are selectively permeable. Some solutes can cross the membrane \_\_\_\_\_, some cross with \_\_\_\_\_, and others do not cross at all.

A few lipophilic (lipid-loving) substances move freely across the cell membrane by \_\_\_\_\_. Most small molecules or ions require the assistance of specific \_\_\_\_\_ to transport them across the membrane. \_\_\_\_\_ molecules do not cross intact cell membranes, except in certain special cases.

### ***Concept 4: Passive and Active Transport***

Most biologically important solutes require \_\_\_\_\_ carriers to cross cell membranes, by a process of either \_\_\_\_\_ or \_\_\_\_\_ transport.

Active transport uses \_\_\_\_\_ to move a solute “uphill” against its gradient, whereas in facilitated diffusion, a solute moves \_\_\_\_\_ its concentration gradient and no \_\_\_\_\_ input is required.

In the illustration below, label which type of transport is shown: *facilitated diffusion* or *active transport*.

